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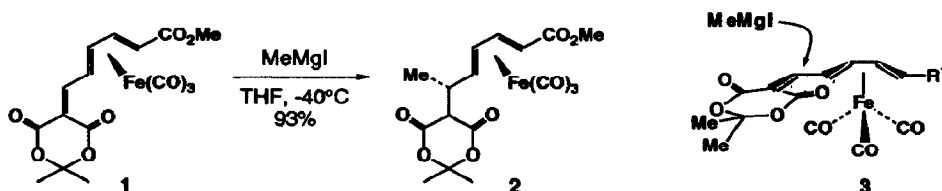
Highly Stereoselective 1,4-Addition Reactions of Alkylidene Malonate Substituted η^4 -(1,3-Butadienyl)iron(tricarbonyl) Complexes

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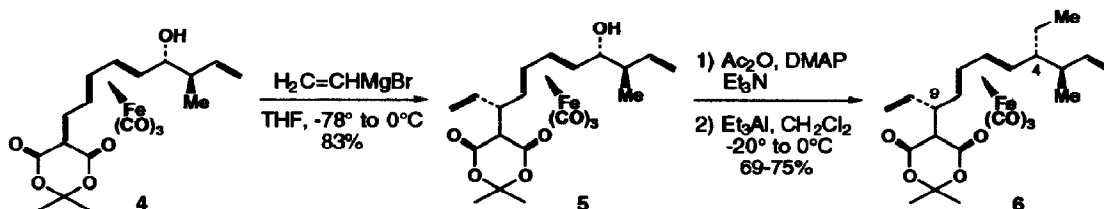
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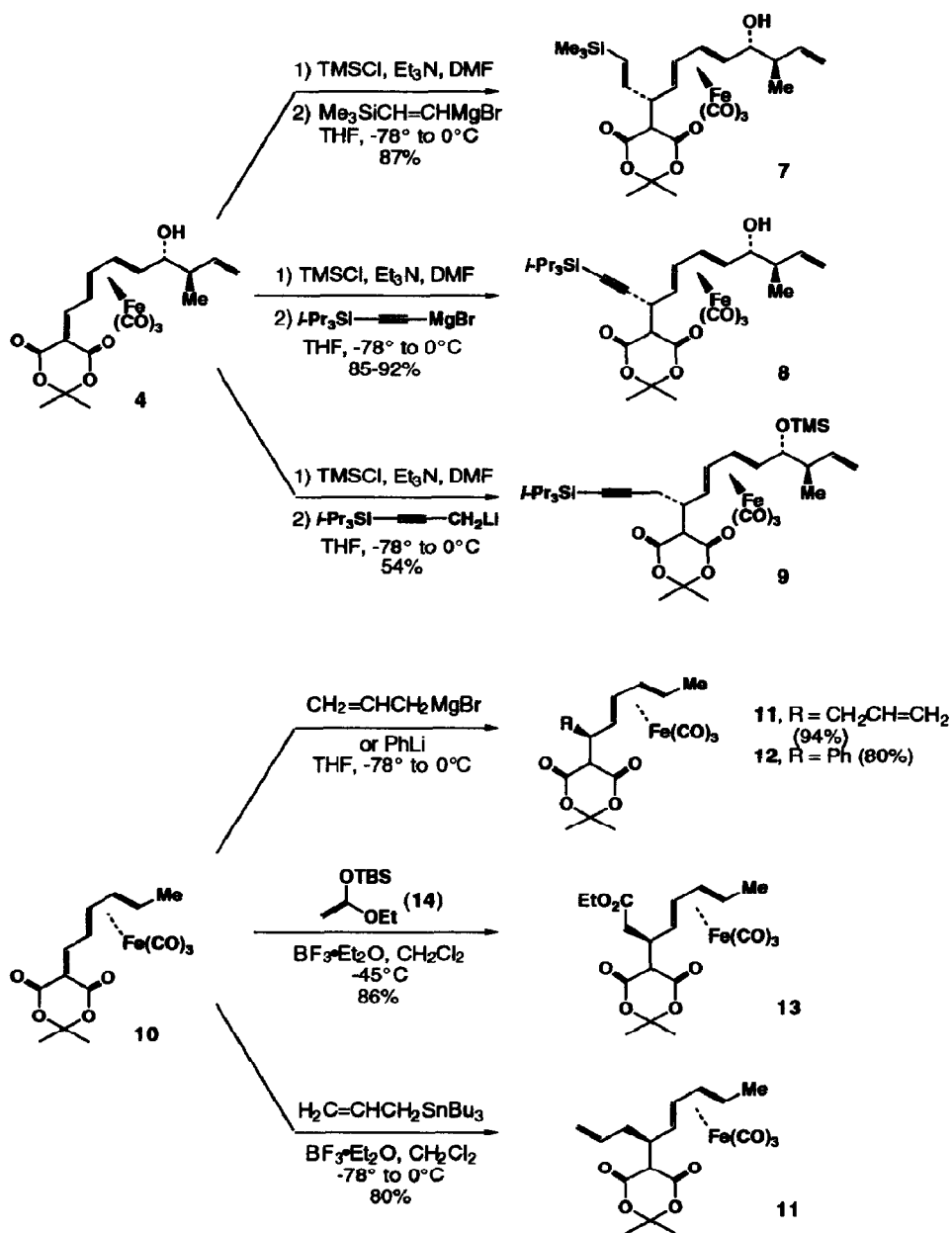
Abstract. Alkylidene malonate substituted η^4 -(1,3-butadienyl)iron(tricarbonyl) complexes **4** and **10** undergo highly diastereoselective 1,4-addition reactions with various nucleophiles, including Grignard reagents, organolithium reagents, ketene silyl acetal **14** and allyltributylstannane.

The conjugate addition of organometallic nucleophiles to alkylidene malonates is well documented in the literature.¹ Grée and Laabassi have demonstrated the highly diastereoselective addition of MeMgI to the η^4 -(1,4-butadienyl)iron(tricarbonyl) substituted alkylidene malonate **1** in the synthesis of (-)-verbenalol and (-)-epiverbenalol.² The -Fe(CO)₃ unit controls the stereochemical course of the 1,4-addition reaction by blocking the bottom face of the alkylidene malonate unit, thereby directing the addition of the carbon nucleophile *anti* to the metal carbonyl substituent (see **3**).³



We have recently employed this conjugate addition methodology in the highly stereoselective synthesis of the *as*-hydrindacene unit of (+)-ikarugamycin.⁴ Specifically, the Meldrum's acid derivative **4** was treated with vinylmagnesium bromide to provide the 1,4-adduct **5** in 83% yield as the only observed stereoisomer (500 MHz ¹H NMR analysis). This reaction coupled with the stereospecific substitution at the carbinol center^{5,6} provided a novel means of controlling the remote stereochemical relationship between C(4) and C(9) of **6**. We report herein studies that define more broadly the scope of the nucleophilic 1,4-addition reactions of η^4 -(1,3-butadienyl)iron(tricarbonyl) substituted alkylidene malonates, using complexes **4** and **10** as representative electrophilic substrates.⁷



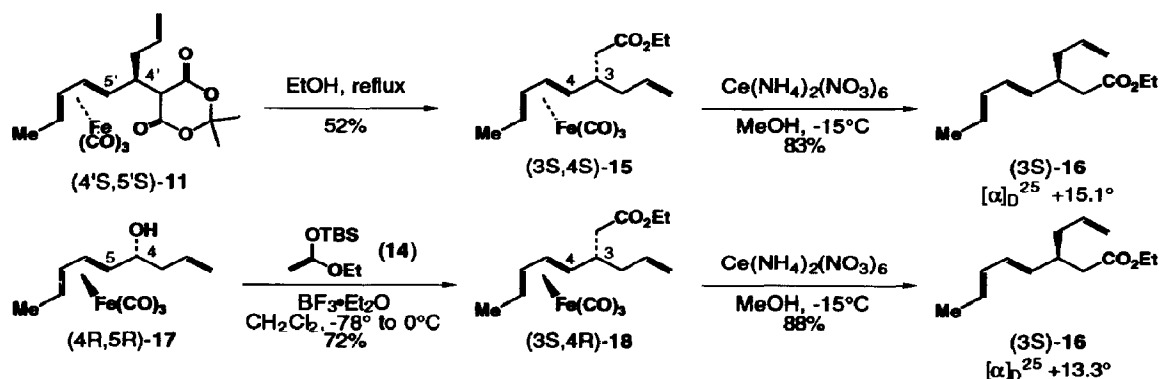


We first examined the reactions of **4** and **10** with several Grignard reagents. The TMS ether of **4** was used in these initial experiments, although our experience with the reaction of **4** and $\text{H}_2\text{C}=\text{CHMgBr}$ suggests that this may not be necessary in all cases.⁴ A -78°C THF solution of the alkylidene malonate complex was treated with a solution of the appropriate Grignard reagent (1.5-2.5 equiv.), and the reaction mixture was allowed to warm slowly

to 0°C. Aqueous saturated NH₄Cl was then added, and the aqueous layer was extracted with Et₂O. The combined organic extracts were washed with saturated aqueous NaHCO₃ and brine, dried over Na₂SO₄, filtered and concentrated by rotary evaporation. The crude products were purified by flash chromatography to provide the 1,4-adducts **7**, **8** and **11** in excellent yield (87-94%).⁸

Thus, alkylidene malonate derivatives **1**,² **4** and **10** undergo highly efficient 1,4-addition reactions with a range of Grignard reagents, including MeMgI, H₂C=CHMgBr,⁴ (E)-Me₃SiCH=CHMgBr,^{9a} and *i*-Pr₃Si-C≡CMgBr.^{9b} Organolithium reagents may also be used, as illustrated by the reaction of **4** with *i*-Pr₃Si-C≡CCH₂Li^{9c} (54% yield of **9**) and of **10** with PhLi (80% yield of **12**). Finally, it is also interesting to note that complex **1** readily undergoes BF₃•Et₂O catalyzed 1,4-addition reactions with ketene silyl acetal **14** (86% yield of **13**) and allyltributylstannane (80% yield of **11**). The alkylidene malonate unit also readily reacts with Et₃Al.¹⁰

A single stereoisomeric 1,4-addition product was produced in each of the reactions summarized herein.¹¹ The stereochemistry indicated in structures **7-9** and **11-13** was assigned by analogy to the precedent provided by the conversion of **1** to **2**,² and by our successful elaboration of **5** to the *as*-hydrindacene nucleus of ikarugamycin.⁴ Additional support for these assignments is provided by the correlation experiment described below. Decarboxylation of the Meldrum's acid unit of (4*S*,5*S*)-**11**¹² by heating in EtOH at reflux provided ethyl ester (3*S*,4*S*)-**15**, which was then treated with Ce(NH₄)₂(NO₃)₆ in MeOH at -15°C provided triene ethyl ester (S)-**16** ([α]_D²⁵ +15.1° (*c* 0.57, CHCl₃)). This compound proved identical to the triene ethyl ester (S)-**16** ([α]_D²⁵ +13.3° (*c* 0.64, CHCl₃)) prepared from (4*R*,5*R*)-**17**⁷ ([α]_D²⁵ +32.8° (*c* 0.40, CHCl₃)) by stereospecific BF₃•Et₂O catalyzed substitution with ketene silyl acetal **14**⁶ and oxidative removal of the -Fe(CO)₃ unit of **18**. Because retention of stereochemistry has been established for substitutions of the type **17** to **18**,⁶ it follows that the stereochemistry of the addition of nucleophiles to the (η⁴-diene)Fe(CO)₃ substituted alkylidene malonates occurs anti to the -Fe(CO)₃ unit, as formulated in transition structure **3**.¹³ It also follows from this analysis that the stereoselectivities of these two processes are complimentary, since the conjugate addition route provides **15** which is a diastereomer of the diene complex **18** produced by the substitution reaction of **17**.

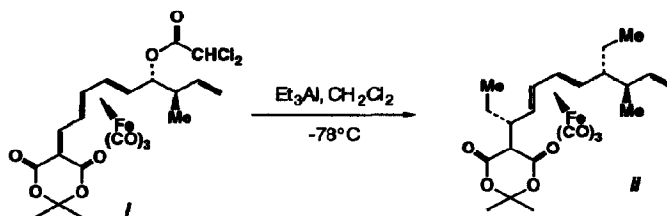


In summary, we have shown that a wide range of carbon nucleophiles undergo highly stereoselective 1,4-addition reactions with η⁴-(diene)Fe(CO)₃ substituted alkylidene malonates of type **4** and **10**. Further applications of the η⁴-(diene)Fe(CO)₃ unit as a stereochemical control element in organic synthesis will be reported in due course.

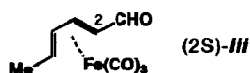
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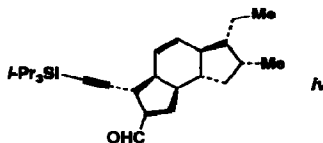
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10. The bis-alkylated product **ii** was obtained as the major product of the reaction of dichloroacetate derivative **i** with Et₃Al at -78°C.



11. Crude reactions mixtures were analyzed by 500 MHz ¹H NMR spectroscopy. Conservatively, the diastereoselectivity of these reactions is $\geq 97 : 3$.
12. (4'S,5'S)-**10** was synthesized starting from the kinetically resolved⁷ sorbaldehyde-Fe(CO)₃ complex (2S)-**iii** ($[\alpha]_D^{25} +103.8^\circ$ (c 0.40, CHCl₃).



13. The stereostructure of **8** also has been verified following its elaboration to **iv** (by using methods similar to those reported in ref. 4). We thank Dr. John. C. Huffmann for performing the X-ray analysis of **iv** (Indiana University Molecular Structure Center Report No. 92316).



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